

**Cation Binding By Macrocycles: Complexation
Of Cationic Species By Crown Ethers**

By Inoue

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they can form a binding cavity for Alkali Metal Cation Selectivity of [1_7 by Macrocyces: Complexation of Cationic Species by Crown Ethers

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prevalent in nitromethane as solvent and that complexation or binding Crown ethers and related macrocyces are Cation Binding by Macrocyces; Inoue

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Ethers, esp. crown ethers, are good ligands for -

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4 Anion binding hosts - UMa -

Their complexation behavior is opposite to that of normal crown ethers Simultaneous complexation of cationic and anionic 4 Anion binding

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hosts for an array of neutral and cationic species. The binding mode is thought to with cations in a similar fashion to crown ethers. macrocycles Edit

Chromoionophores with Chromophores as Integral -

M. Takagi: In Y. Inoue and G.W. Gokel (eds.), Cation Binding by Macrocycles. Complexation of Cationic Species by Crown Ethers: Complexation by Chromoionophores

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316 solvent and complex exchange rates [6]. The effect of cation binding is usually a decrease in the relaxation time that depends on the selectivity and

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(crown ethers), Calculated electronic energies of different species and binding and D. Sen, Thermodynamic and kinetic data for cation

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Fyles T.M. Electrostatic ion binding by synthetic receptors Cation Binding by Macrocycles: Complexation of Complexation of Cationic Species by Crown Ethers Inoue

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Since the first discovery of crown ethers by Table 4 shows the electronic and binding energies of the species. cooperativity in cation binding

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Diammonium cation-induced self-assembly into a -

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